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# A REVIEW OF THE ANTIPARASITIC EFFECTS OF ELLAGIC ACID AND OTHER PHENOLIC COMPOUNDS ISOLATED FROM MEDICINAL PLANTS

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#### **REVIEW ARTICLE**

History	Abstract
Received: 19 <sup>th</sup> March 2023	The pursuit of alternative sources for antiparasitic drugs is crucial due to the frequent
Accepted: 8 <sup>th</sup> July 2023	inadequacy of current treatments and the potential development of resistance among
	parasites towards synthetic therapies. Conversely, medicinal plants have garnered
Keywords:	prominent attention as it produces various natural compounds with intriguing biological
Ellagic acid; Phenolic compounds; Plasmodium; Trypanosoma; Leishmania; Medicinal plants	properties that can be beneficial to treat parasitic infections. This article presents a comprehensive overview of past literature regarding the topic. It specifically focuses on the antiparasitic effects of ellagic acid and other phenolic compounds against parasites, such as <i>Plasmodium</i> , <i>Trypanosoma</i> , <i>Leishmania</i> , <i>and Helminths</i> . The findings hope to contribute further understanding on the utilisation of ellagic acid and other phenolic compounds as innovative antiparasitic drugs for effectively managing parasitic infections. It also highlights potential limitations and set the direction for future research.

## INTRODUCTION

The global population has witnessed a widespread of parasitic diseases with approximately 1 to 2 billion infection cases and several million deaths being reported annually [1]. It is commonly caused by parasitic infections with notable examples include malaria, trypanosomiasis, leishmaniasis, and helminthiases [2]. Many antiparasitic drugs have been introduced as a countermeasure to such issue [3]; however, certain parasite strains have developed resistance to these drugs over time. This leads to the lack of effective drugs, subsequently posing a challenge for the prevention of parasite infections [2].

Medicinal plants have been traditionally used to treat parasitic infections. Past literature suggests that efforts to initiate the search for antiparasitic natural compounds should begin in Asia, Africa, and Latin America [2] as these nations have a greater number of publications on traditional medicine and possess a larger variety of medicinal plants compared to other nations [4]. It is feasible to discover novel bioactive components as medicinal plants abound in natural compounds with a wide range of structural diversity [5]. Ellagic acid and other phenolic compounds derived from plants have undergone frequent laboratory examinations [6, 7, 8] and are believed to hold the potential for antiparasitic actions [9, 10, 11].

Crude extracts from various parts of medicinal plants, such as *Punica granatum* roots and *Carica papaya* roots, leaves, and seeds, are believed to possess efficacy against various helminths [12]. Compounds like benzyl thiocyanate isolated from *C. papaya* seeds and diospyrol from *Diospyros mollis* fruits have demonstrated effectiveness against *Ascaris lumbricoides* and *Necator americanus*, respectively [12]. Extracts of *Lippia multiflora*, *Aframomum sceptrum*, and *Uvaria afzelii* have exhibited antileishmanial activity (IC<sub>50</sub> =  $<25 \mu g/mL$ ) [13]. The reference drug, miltefosine, demonstrated antimalarial activity with an IC<sub>50</sub> of 8.219 ± 0.6337 µg/mL [14]. In Ivory Coast, *Lawsonia inermis* is specifically prescribed for the treatment of African trypanosomiasis [8]; however, no efficacy has been

demonstrated against other parasites. Extracts from *Terminalia glaucescens* and *Anogeissus leiocarpus* have also exhibited antimalarial activity ( $IC_{50} = <5 \ \mu g/mL$ ) and are recommended by traditional healers for the prevention of malaria infections [13].

### PHENOLIC COMPOUNDS

The primary products of plant secondary metabolism are phenolic compounds. Apart from protecting plants against parasites and insects, phenolic compounds are responsible for pigmentation, astringency, and UV light protection [15]. These substances are present in a wide variety of matrices, including fruits (blueberries and apples) and vegetables (potatoes and red cabbage) [16, 17]. In terms of molecular characteristics, phenolic compounds are characterised by one or more aromatic rings bonded to one or more hydroxyl substituents, resulting in diverse chemical structures [18]. These compounds are categorised into five groups: flavonoids, tannins, lignans, stilbenes, and phenolic acids.

#### Flavonoids

As shown in Figure 1A, flavonoids can be categorised into numerous subgroups based on their unsaturation and oxidation status (the carbon of the C ring is bonded to the B ring) [19]. In the case of isoflavones, the B ring is attached to position 3 of the C ring, while for neoflavonoids, the B ring is linked to position 4 of the C ring (Figure 1B-C). A variety of flavonoid subgroups have their B rings attached at position 2 of the C rings, including flavones, flavonols, flavanones, flavanools, flavanols or catechins (flavan-3ols), anthocyanins, and chalcones (Figure 1D-J) [19].

Several flavonoids have been extensively investigated and have demonstrated activity within the concentration range of 5-32 µM against Cryptosporidium parvum and Entamoeba intestinalis [20]. Specific flavonoid compounds, such as apigenin and genistein, are known as protein tyrosine kinase inhibitors and have exhibited inhibitory effects against both C. parvum and Toxoplasma gondii [21, 22]. Earlier studies reported that genistein, a protein tyrosine kinase inhibitor, impedes calcium ionophores from inducing the release of Toxoplasma gondii tachyzoites from host cells, including neutrophils or kidney cells [23]. Apigenin exerts its inhibitory action against C. parvum through the induction of host cell apoptosis [20]. Phenylpropanoid and Chromone flavonoids have demonstrated effectiveness against both chloroquine-sensitive and chloroquine-resistant strains of Plasmodium falciparum in vitro [24]. Additionally, flavonoids such as casticin and chrysosplentin are believed to hinder the production of fatty acids by parasites [25, 26, 27]. Notably, flavonoids have been identified as potent active agents in an in vitro assay targeting Leishmania amazonensis amastigotes. Among these, the flavonoid compound kaempferol 3,7-di-O-methyl ether exhibited significant efficacy against amastigotes, yielding the IC<sub>50</sub>

value of  $10.5 \pm 2.5 \mu$ M. Comparatively, previous research indicates that the commercial drug miltefosine holds the IC<sub>50</sub> value of  $18.933 \pm 0.737 \,\mu$ M for the *Leishmania amazonensis* CU1 form, with effective IC<sub>50</sub> values ranging from 3.88 to 28.7 (>306.6) uM [28, 29, 30]. Moreover, a distinct study revealed that quercetin, gallic acid, and rutin compounds also exerted effectiveness against L. donovani, yielding the IC<sub>50</sub> values of 84.65 µg/mL, 86 µg/mL, and 98 µg/mL, respectively [31]. These compounds induced alterations in the ultrastructure of treated cells, ultimately leading to parasite demise. Further effects included the enhancement of nitric oxide generation, initiation of apoptosis, and DNA damage. Additionally, these compounds suppressed critical enzymes trypanothione reductase and trypanothione synthetase that are essential for leishmanial survival [31]. In a previous investigation, phenolic and flavonoid compounds within the plant extracts of *P. retrofractum* (IC<sub>50</sub> =  $87.63 \pm$ 3.55  $\mu$ g/mL), A. gangetica (IC<sub>50</sub> = 39.06 ± 1.98  $\mu$ g/mL), and A. scholaris (IC<sub>50</sub> = 48.92  $\pm$  1.52 µg/mL) demonstrated pronounced inhibitory activity against the proteolytic subunit of the caseinolytic protease system of P. knowlesi (Pk-ClpP) [32].

#### Tannins

Tannins refer to flavonoid polymers that can be classified into two subclasses: condensed tannins and hydrolysable tannins. Mixtures of simple phenols can give rise to hydrolysable tannins (Figure 2) [33]. Under the influence of acidic or basic conditions, these compounds can undergo hydrolysis, leading to the formation of phenolic acids and carbohydrate molecules. For example, gallic acid can serve as a precursor for hydrolysable tannins (Figure 2A). Condensed tannins are assembled through the linkage of two or more monomers of flavan-3-ol units, primarily connected via bonds between their A rings and the pyran rings of other flavanols (Figure 2B). Due to their intricate structural nature, tannins have not been extensively explored in terms of their bioactive potential [18]. However, proanthocyanidins have recently garnered attention due to their potential health benefits, including anti-inflammatory and antimicrobial activities [18, 34].

Distinct variations in activity have also been observed when goats were administered with condensed tannins before being exposed to third-stage larvae, resulting in reductions of worm populations by 33%, 70%, and 66% for *H. contortus, T. circumcincta, and T. colubriformis,* respectively [35]. Condensed tannins exhibit chelation properties for iron and zinc that are crucial for the growth and reproduction of *P. falciparum.* While the role of iron is better understood, the function of zinc is less recognised. The parasite-driven influx of poorly bound zinc into infected erythrocytes contributes to the parasite's pathogenicity. Prior to host cell rupture, the *P. falciparum* parasite must acquire 30 million zinc atoms, corresponding to a 400% increase in overall zinc concentration. Notably, highly zinc-specific chelators have demonstrated the capability to arrest the parasite's development at the trophozoite stage while concurrently inhibiting parasite proliferation. The presence of sufficient chelator concentration renders the parasite nonviable [36]. Furthermore, the tannin-treated larvae of *A. suum* displayed notable effects with all larvae succumbing to

treatment at a concentration of 1 mg/mL. Even at a concentration of 111 mg/mL, motility was greatly diminished. These findings strongly suggest that condensed tannins exert potent anthelmintic effects on both the L4 and L3 stages of *A. suum* [37].



Figure 1. The basic chemical structure of flavonoids and their classes.



Proanthocyanidins B (Interflavanic bond C8-C6)

Figure 2. The chemical structure of tannins. (A) Hydrolysable tannins, which are allotannins and ellagitannins. (B) Condensed tannins, which are proanthocyanidins A, proanthocyanidins B and proanthocyanidins C.

#### Lignans

The fundamental structure of lignans consists of an amalgamation of phenylpropanoid dimers (C6-C3) that are interconnected through the central carbons of their side chains, as shown in Figure 3 [38]. Lignans represent secondary plant metabolites exhibiting diverse chemical configurations that are often found in specific seeds. It also exhibits both estrogenic and antiestrogenic properties than various other bioactivities [38]. Among the isolated tetrahydrofuran lignans, calopeptin demonstrated activity against *P. falciparum* D6 and W2 strains, yielding the IC<sub>50</sub> values of 3800 ng/mL and 3900 ng/mL, respectively [39].

Notably, lignans machilin-G and veraguensin exhibited promising antileishmanial efficacy against *L. donovani*, yielding the IC<sub>50</sub> values of 18 µg/mL and 36 µg/mL, respectively [39]. In terms of potency against *T. cruzi* trypomastigotes, both copalic acid and kusunokinin emerged as the most effective lignans when compared to fargesin, epieudesmin, and sesamin lignans. The reference drugs employed in the study included pentamidine, glucantim, and benznidazole [40]. Copalic acid displayed selectivity against parasitized erythrocytes and minimal toxicity towards other mammalian cells. Notably, kusunokinin showed substantial efficacy against trypomastigotes (IC<sub>50</sub> = 51 µM) without inducing hemolytic activity [40].



Lightin

Figure 3. The chemical structure of lignans.

#### Stilbenes

The chemical structure of stilbenes, characterised by the presence of a double bond linking two benzene rings, is represented by the formula C6-C2-C6 (Figure 4 A) [18]. Stilbenes can be categorised into Z and E isomers based on the location of their central double bond. While over 400 stilbenes have been identified, only plant families possessing the pivotal enzyme necessary for stilbene biosynthesis can produce these compounds. Among the prominent stilbenes, resveratrol stands out (Figure 4 B) and has been extensively researched due to its notable bioactivities [18]. Stilbenebased compounds have demonstrated antiplasmodial action, inducing stage-specific apoptotic death in young Plasmodium parasites [41]. The occurrence of DNA fragmentation and loss of mitochondrial membrane potential serves as evidence of the parasite-specific programmed cell death initiated by stilbene-based compounds. Furthermore, stilbene ST18 has been reported to become active against T.

*cruzi* amastigotes and exhibits lower cytotoxicity when evaluated in vitro on normal healthy cells [42]. It is theorised that stilbene ST18 activates the conserved enzyme caspase-1 in infected macrophages, which is an essential regulator of parasitemia, host survival, and the onset of the adaptive immune response in Trypanosoma infection.

#### **Phenolic Acids**

Plant phenolics exhibit numerous similarities with aliphatic alcohols [43]. Phenolics function as weak acids rather than strong alcohols due to their possession of an aromatic ring and a hydrogen atom within the hydroxyl group. These compounds are crucial for performing various bioactivities, including supporting plant development, growth, and defense, while also exerting positive health impacts on humans [44]. Phenolics are recognised natural antioxidants that play a significant role in various biological and pharmacological activities, such as anti-inflammatory, anticancer, antimicrobial, antiallergenic, antiviral, antithrombotic, hepatoprotective, and antiparasitic [45, 46, 47, 48]. Two hydroxybenzoic acids (Figure 5) and five hydroxycinnamic acids (Figure 6) have been evaluated for their inhibitory potential against the growth of *P. falciparum* (3D7 clone) [49]. Both protocatechuic acid and gallic acid displayed the IC<sub>50</sub> values of 30.08 µg/mL and 17.73 µg/mL, respectively. Additionally, caffeic acid, o-coumaric acid, chlorogenic acid, ferulic acid, and rosmarinic acid (Figure 6A-E) yielded the IC<sub>50</sub> values of 26.59 µg/mL, 82.23 µg/mL, 105.76 µg/mL, 93.36 µg/mL, and 103.59 µg/mL, respectively [49].

Rosmarinic acid is proven to be more effective in dismantling the intracellular promastigote and amastigote

forms of *L. donovani* [50]. The mechanism of action involves iron chelation, resulting in cell cycle arrest and various morphological alterations [50]. Past research investigated gallic acid in an anthelmintic bioassay to determine the time of paralysis and death in the earthworm *Pheretima posthuma*. It was found that the worm exposed to gallic acid exhibited dose-dependent paralysis and eventual demise [51]. Furthermore, gallic acid was tested for its antitrypanosomal activity against *T. brucei brucei* in vitro (IC<sub>50</sub> = 14.2 ± 1.5  $\mu$ M). This compound influenced both mitochondrial membrane integrity and cell viability in a dose-dependent manner. Moreover, gallic acid-induced kinetoplast loss in trypanosomes [52].



Figure 4. The chemical structure of (A) stilbene and (B) resveratrol.



Figure 5. The chemical structure of hydroxybenzoic acids, which are (A) protocatechuic acid and (B) gallic acid.



Figure 6. The chemical structure of hydroxycinnamic acids, which are (A) caffeic acid, (B) o-coumaric acid, (C) chlorogenic acid, (D) ferulic acid and (E) rosmarinic acid.

### ELLAGIC ACID

Ellagic acid is a natural phenolic constituent with a molecular mass of 302.197 g/mol (Figure 7). It is an amphiphilic molecule consisting of a planar biphenyl lipophilic moiety joined by two rings and possesses four

hydroxyl groups that are combined with lactone groups to form a hydrophilic moiety, which plays a crucial role in the compound's biological activity [53, 54, 55]. Currently, ellagic acid has garnered significant attention due to its remarkable antiparasitic effects [56, 57, 58].



Ellagic acid

Figure 7. The chemical structure of ellagic acid.

#### Antiparasitic Activities of Ellagic Acid

A group of researchers from the University of Milan made the initial discovery of ellagic acid isolated from Tristaniopsis calobuxus [59] and Punica granatum in Malaysia [60]. They found that this compound inhibited the growth of chloroquine-sensitive and chloroquine-resistant strains of P. falciparum in vitro [59]. Furthermore, ellagic acid's activity was found to be more pronounced during the mature stages of the parasite's lifecycle when the majority of the haemoglobin-rich host cell cytoplasm was consumed and digested [59]. One proposed mechanism of ellagic acid's antimalarial action is the suppression of beta-hematin (hemozoin) formation, a process by which the parasite neutralises the toxic by-products of haemoglobin breakdown [59]. In a recent study by Muchtar et al. (2022) [61], ellagic acid inhibited P. falciparum (3D7 strain) with an IC50 of 1.85  $\pm$  4.57 nM. A significant change was further observed in the pH of the digestive vacuole of ellagic acid-treated parasites (pH values ranging between 6.11 to 6.74) compared to nontreated parasites (pH 5.18). Another investigation by Haeussler et al. (2019) [62] demonstrated that ellagic acid substantially inhibited P. vivax glucose 6-phosphate dehydrogenase (PvG6PD) with an IC<sub>50</sub> of  $32.5 \pm 13.4$  nM and P. falciparum glucose 6-phosphate dehydrogenase 6phosphogluconolactonase (PfGluPho) with an IC<sub>50</sub> of 77  $\pm$  22 nM.

Ellagic acid was also reported to exhibit substantial antimalarial activity in an intraperitoneal administration study of a murine model [59]. This compound showed greater efficacy against P. vinckei petteri compared to artesunate. Mice treated with ellagic acid before parasite inoculation displayed a greater reduction in parasitemia than control mice, suggesting a therapeutic effect of the compound. Ellagic acid also demonstrated synergistic effects with chloroquine (FIC<sub>50</sub> =  $0.63 \pm 0.05$ ), mefloquine  $(FIC_{50} = 0.64 \pm 0.09)$ , atovaquone  $(FIC_{50} = 0.53 \pm 0.21)$ , and artesunate (FIC<sub>50</sub> = 57  $\pm$  0.13), resulting in lower concentrations of each drug used in combination therapy and potentially preventing drug resistance [59]. To assess its effect on chloroquine-resistant reversion in the P. falciparum W2 strain, ellagic acid was compared with verapamil, which is known to reverse chloroquine resistance. The study concluded that verapamil exhibited chloroquine resistance reversal while ellagic acid did not. Additionally, chloroquine-sensitive strain F32 served as a negative control in the study and did not show such reversion effect [59].

Past research also reported the antileishmanial effect of ellagic acid (maximal effect =  $100 \ \mu g/mL$ ) on *L. major* promastigote forms. Results from the MTT test revealed that

ellagic acid decreased sheep macrophage viability starting at 6.3 µg/mL, with a CC<sub>50</sub> value of 23.8 µg/mL [63]. Another investigation demonstrated that ellagic acid displayed strong antileishmanial activity (IC<sub>50</sub> = 151.7 µg/mL) against the promastigotes of *L. amazonensis*. Treatment with ellagic acid further resulted in a decreased percentage of amastigotes in infected macrophages [64].

Moreover, the effectiveness of ellagic acid against the cattle helminth *Haemonchus contortus* was evaluated using the adult motility and egg hatch tests in vitro. In the adult motility test, ellagic acid (concentrations ranging from 0.09-3.00 mg/mL) inhibited *H. contortus* in a concentration-dependent manner. Test worms exposed to ellagic acid at 1.5 mg/mL died six hours after exposure. In the egg hatch test, ellagic acid (concentrations ranging from 0.0125-25.00 µg/mL) demonstrated a concentration-dependent reduction in larval generation from *H. contortus eggs* [65].

In another study, rats infected with *Trypanosoma* congolense received daily ellagic acid treatments of 100 and 200 mg/kg body weight (BW) for 14 days. This compound exhibited significant trypanosuppressive activity at doses of 100 and 200 mg/kg (p < 0.05). Furthermore, ellagic acid considerably reduced trypanosome-induced anaemia, hepatic and renal damage, hepatomegaly, splenomegaly, and renal hypertrophy (p < 0.05) [66].

The mentioned  $IC_{50}$  values are promising for further in vivo testing and drug development. Analogue research and iterative medicinal chemistry have been employed to identify compounds with enhanced efficacy and pharmacological characteristics. Understanding the target molecule's structure can effectively guide medicinal chemistry efforts. Reliable, highly predictive in vitro and in vivo testing for activity, often using the same parasite species that infect humans, is advantageous for new antiparasitic drug development [67].

Lead optimisation involves designing and synthesising novel compounds using medicinal chemistry, followed by evaluating their improved properties. This iterative process impacts costs and timelines, with the lead optimisation stage being pivotal in shaping the drug discovery process [67]. Selecting a strong drug candidate at this stage involves prominent consideration of bioavailability, metabolic factors, and toxicity. It essentially involves problem-solving where the involvement of the pharmaceutical sector becomes crucial. In recent years, public-private partnerships have significantly improved the lead optimisation phase of drug discovery by re-engaging the antiparasitic pharmaceutical industry and the provision of funds. These partnerships have achieved promising results in getting drug candidates into clinical trials and addressing how businesses can contribute to the cause. Public-private partnerships have also adopted portfolio strategies that often prioritise development over discovery, despite the inherent risks in drug discovery. While government funding supports fundamental and early discovery research, collaborations between public entities in early discovery research are also viable [67].

## LIMITATION OF THE NATURAL COMPOUND-BASED ANTIPARASITIC DRUG DISCOVERY

Natural product research involves numerous practical considerations, many of which become evident upon exploring the subject. These concerns encompass the reliability of indigenous medicinal data, the availability of sufficient supplies of the resource, and the potential risks of overharvesting and resource depletion [68]. Factors such as plant subspecies or age, regional and seasonal variations, extraction and storage methods, as well as climatic influences can collectively influence the quality and consistency of chemical constituents and their concentrations in natural products [69]. The isolation of pure compounds from chemically complex extracts, often used for screening, can also pose significant challenges [12]. The utilisation of local medicinal herbs may yield valuable insights into lesser-known plants; however, different communities might have diverse regional applications for the same plants. To potentially address these challenges, the establishment of smaller, specialised organizations focusing on natural product drug discovery can facilitate the licensing of their discoveries to larger entities for marketing and clinical development. An example of successful collaboration between a smaller organisation and a major pharmaceutical company is Pharmamar [71].

# CONCLUSION

Plants produce an abundance of natural compounds with intriguing biological characteristics. The potential to uncover novel bioactive components with antiparasitic properties is conceivable due to the presence of natural chemicals with diverse structural variations in medicinal plants. Ellagic acid and other phenolic substances are routinely analysed in laboratories. Prior studies have demonstrated the potential antiparasitic effects of phenolic compounds, including the potential for ellagic acid to inhibit the development of parasites such as Plasmodium, Trypanosoma, Leishmania, and helminths. Further studies should be conducted to comprehensively investigate the properties of ellagic acid and other phenolic compounds.

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#### **CONFLICT OF INTEREST**

The authors declare that there is no conflict of interest regarding the publication of this manuscript.

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